

This article was downloaded by:

On: 28 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

## Further Adventures in Germanium NMR

Ralph A. Thomson; Alistair L. Wilkins; Kenneth M. Mackay

**To cite this Article** Thomson, Ralph A. , Wilkins, Alistair L. and Mackay, Kenneth M.(1999) 'Further Adventures in Germanium NMR', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 150: 1, 319 — 324

**To link to this Article:** DOI: 10.1080/10426509908546400

**URL:** <http://dx.doi.org/10.1080/10426509908546400>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## Further Adventures in Germanium NMR

RALPH A. THOMSON\*, ALISTAIR L. WILKINS and KENNETH  
M. MACKAY

*Chemistry Dept., University of Waikato, P.B. 3105, Hamilton, N.Z.*

The acoustic ring signal that dominates NMR spectra of low gyromagnetic ratio nuclei has always been a problem when trying to observe signals from weak samples. This work looks at some ways of improving the observability of  $^{73}\text{Ge}$  – one such nucleus.

**Keywords:** Germanium; NMR; Baseline Roll

### INTRODUCTION

During the 1980's the earlier pioneering observations on  $^{73}\text{Ge}$  NMR were followed up<sup>[1]</sup> by the groups of Takeuchi, Lukevics and ourselves largely using an observing frequency of 3.1 MHz. Compounds observed included hydrides, alkylgermanes, alkylpolygermanes, silicon-germanium hydrides and alkyl derivatives, tetraalkyl germanes, carbofunctional alkyls, tetraalkoxides, and tetrahalides.

The work highlighted the problems of working with high spin, low resonance frequency nuclei. The baseline roll resulting from 'acoustic ringing' creates difficulties in trying to observe the expected broader line

---

\* *r.thomson @waikato.ac.nz*

signals from non-symmetric species within wide spectral windows.

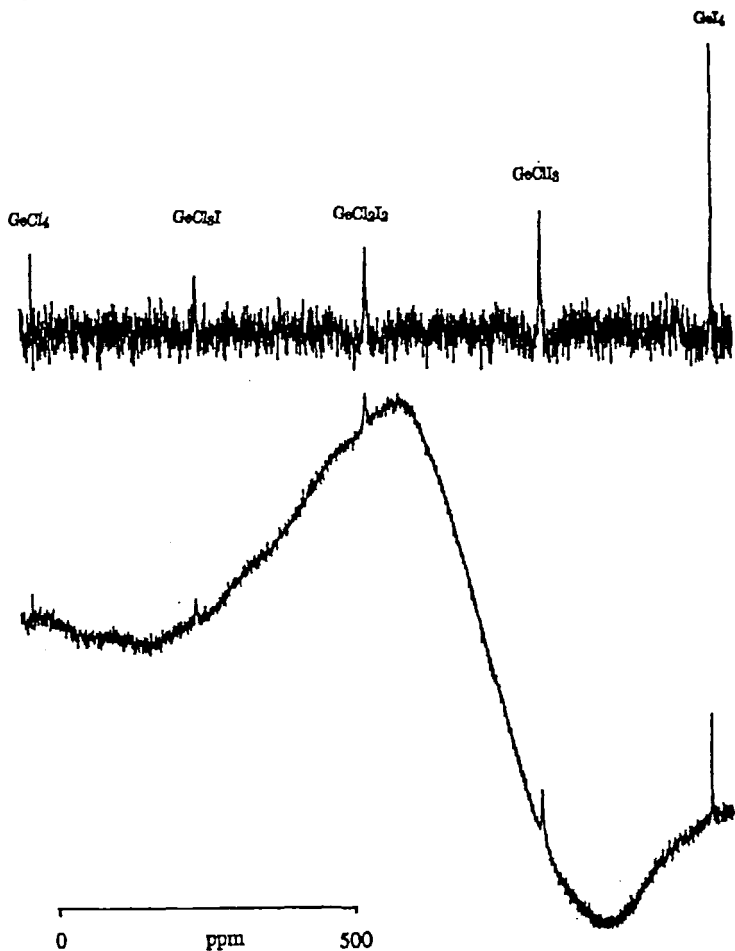


FIGURE 1  $^{73}\text{Ge}$  spectra, at 10.5 MHz, of a mixture of  $\text{Me}_2\text{GeCl}_2$  and  $\text{GeI}_4$  (in excess). See text for spectral details

The common approach is to zero the first few data points of the FID. For example Figure 1 shows spectra resulting from the mixing of  $\text{Me}_2\text{GeCl}_2$  and  $\text{GeI}_4$ . The lower trace is a standard one pulse experiment, while the upper trace shows the reasonably smooth baselines

obtained by zeroing 32 data points. However, any broad-line, fast relaxing species would be lost. The spectra show only the mixed tetrahalide species, with no  $^{73}\text{Ge}$  signals from  $\text{Me}_2\text{GeCl}_2$ ,  $\text{Me}_2\text{GeClI}$  or  $\text{Me}_2\text{GeI}_2$  although  $^{13}\text{C}$  and  $^1\text{H}$  NMR showed that these species were present.

To observe broadline species a number of pulse sequences have been used, including RIDE and EXSPEC<sup>[2]</sup>.

In recent times only occasional observations of germanium have been reported<sup>[3]</sup>. The studies to date have shown that there are straight-line  $^{73}\text{Ge}$ ,  $^{29}\text{Si}$ , and  $^{73}\text{Ge}$ ,  $^{119}\text{Sn}$  chemical shift correlations between equivalent germanium, silicon, and tin, compounds<sup>[1]</sup>.

Observations made to date raise some questions. While mixed tetra-alkyl and mixed tetra-halide species have been observed no mixed  $\text{R}_n\text{GeX}_{4-n}$  species have been observed (see Figure 1). As mixed halide species such as  $\text{GeCl}_2\text{I}_2$  are expected to have greater field gradients than some of the mixed alkyl halide species, why are the former, but not the latter observed?

Even with such pulse sequences the broadest signal observed to date is 500 Hz, considerably narrower than signals observed in other multinuclear NMR experiments of quadrupolar nuclei. Why is it that broader lined species are not observed?

## CURRENT WORK

We are presently looking at what improvement higher field (300 MHz), and a dedicated germanium probe gives to the observation of this low resonance frequency nucleus. Higher field NMR improves the observation frequency and the sensitivity, but to maintain reasonable observation windows more powerful pulses are needed, and this aggravates the 'acoustic ring'. The current generation of spectrometers allow for greater use of composite pulses, and greater phase cycling. This assists in providing better suppression of rolling baselines using the

various pulse sequences available. Figure 2 shows what can be achieved using the more detailed pulse sequences.

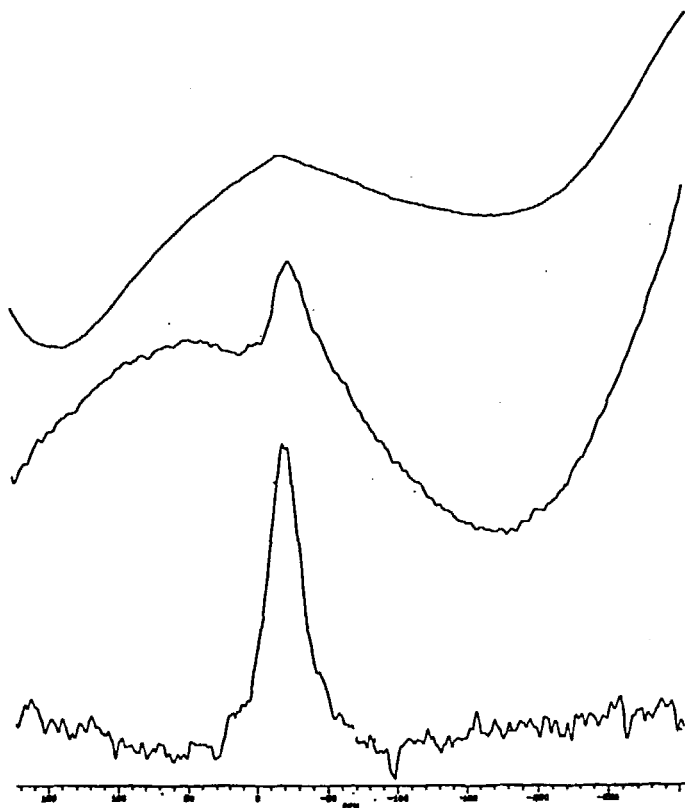


FIGURE 2  $^{73}\text{Ge}$  NMR spectrum of  $\text{EtGe}(\text{Ph})_3$ . The upper trace is a standard one pulse experiment and shows no obvious signal. The middle trace was obtained using the RIDE sequence and while showing a signal from the sample still contains considerable baseline roll. The lower trace was obtained using the EXSPEC sequence and shows very good removal of the rolling baseline. Each spectrum was acquired over 8 hrs.

Table 1 shows recent, previously unpublished observations, including those made at higher field.

TABLE 1 Recent Germanium-73 Chemical Shift Observations.

Compound	Chemical Shift (ppm) <sup>(a)</sup>	Linewidth (Hz)
Observations made at 3.1 MHz.		
GeH <sub>3</sub> GeMe <sub>2</sub> Cl	-280.5 (only GeH <sub>3</sub> observed)	39
GeH <sub>3</sub> GeH <sub>2</sub> Mn(CO) <sub>5</sub>	-291.8 (only GeH <sub>3</sub> observed)	68
GeH <sub>3</sub> GeH(Me)Mn(CO) <sub>5</sub>	-277.9 (only GeH <sub>3</sub> observed)	17
GeH <sub>3</sub> D	-293.3	1.95
GeD <sub>4</sub>	-299.1	3.1
Ge <sub>2</sub> D <sub>6</sub>	-318.0	63
Observations made at 10.5 MHz.		
Ge <sub>2</sub> Ph <sub>6</sub>	-67	90
Ge <sub>2</sub> Me <sub>6</sub>	-59	
MeGePh <sub>3</sub>	-22.9	43
EtGePh <sub>3</sub>	-19.7	
(Ph <sub>3</sub> Ge) <sub>3</sub> GeH	-314	200
(Ph <sub>3</sub> Ge) <sub>3</sub> SiH	-53.2	40
(Ph <sub>3</sub> Sn) <sub>4</sub> Ge	-480.4	24
(a) w.r.t. Me <sub>4</sub> Ge at 0 ppm.		

The availability of computer programs for testing the theoretical efficiency of pulse sequences, and composite pulses, allow the non-mathematical among us to more easily carry out tests on likely sequences

without tying up too much spectrometer time. Work is continuing on efforts to improve the observation windows.

Germanium NMR is a useful tool in looking at exchange reactions such as those shown in Figure 1. Proton or carbon NMR, for other than simple R groups, may contain far too much overlap to obtain useful kinetic data from this type of reaction. We are working on a number of halide exchange reactions.

To better understand the effect substituents have on the field gradient, and therefore the line widths, there is a need to look at a greater range of germanium compounds.

### Experimental

A sample of  $\text{Me}_2\text{GeCl}_2$  ( $^1\text{H}$  0.8 ppm/ $^{13}\text{C}$  11.1 ppm), containing 8% of  $\text{MeGeCl}_3$  ( $^1\text{H}$  1.3 ppm/ $^{13}\text{C}$  16.4 ppm), was dissolved in  $\text{CS}_2$  and an excess of  $\text{GeI}_4$  added. At equilibrium a signal for  $\text{Me}_2\text{GeCl}_2$  ( $^1\text{H}$  1.2 ppm/ $^{13}\text{C}$  14.4 ppm) containing approx 5% of the total Cl was observed. Peaks for the other species in solution totalled less than 1%.

### Acknowledgments:

We thank Professor D.K. Breiting and Manfred Krüger, Institut für Anorganische Chemie der Universität Erlangen-Nürnberg for the supply of compounds used in the observations at 10.5 MHz.

### References

- [1] K.M. Mackay and R.A. Thomson, *Main Group Metal Chemistry* **10**, 85 (1987).
- [2] A.L. Wilkins, R.A. Thomson and K.M. Mackay, *Main Group Metal Chemistry* **13**, 219, (1990).
- [3] (i) Y. Takeuchi, K. Ogawa, G. Manuel, R. Boukherroub and I. Zicmane, *Main Group Metal Chemistry* **17**, 121, (1994). (ii) Y. Takeuchi, I. Zicmane, G. Manuel, R. Boukherroub, *Bull. Chem. Soc. Jpn* **66**, 1732, (1993).